

CREST Science Olympiad (CSO)
Worksheet

Class 6

Topic

Separation of Substances









Worksheet on Separation of Substances

1. Which of the following statements describes the relationship between a pure substance and an element correctly?

- a. A pure substance and an element are the same thing.
- b. A pure substance is always an element.
- c. An element is a mixture of pure substances.
- d. While all elements are pure substances, not all pure substances are elements.

2. In a laboratory experiment, a mixture of iron filings and sulfur powder is heated strongly. What is the expected outcome of this experiment?

- a. The iron filings will dissolve in the sulfur powder.
- b. The sulfur powder will sublime, leaving the iron filings behind.
- c. The iron filings will react with the sulfur powder to form a new compound.
- d. The sulfur powder will melt, while the iron filings remain unchanged.

3. Among the given substances, which ones are compounds?

- a. Iron, lemonade, oxygen, steel
- b. Sugar, glass, carbon dioxide, oxygen gas
- c. Paint, gold, milk, soil
- d. Wood, ink, vinegar, bronze

4. What is the correct order for obtaining ethanol from a mixture of ethanol and water?

- a. Boiling Condensation Collection
- b. Condensation Boiling Collection
- c. Collection Distillation Boiling
- d. Boiling Collection -Condensation

Olympiads

5. Match the following methods of separation with their corresponding descriptions.

	Column I		Column II
1.	Filtration	A.	Separation of grains from stalks by beating or crushing
2.	Magnetic separation	B.	Separation of solid particles from a liquid or gas by passing it through a porous medium
3.	Winnowing	C.	Separation of magnetic substances from non-magnetic substances
4.	Threshing	D.	Separation of heavier or lighter components of a mixture by wind or blowing air

- a. 1:B, 2:C, 3:A, 4:D
- b. 1:C, 2:A, 3:D, 4:B
- c. 1D:, 2:C, 3:A, 4:B
- d. 1:B, 2:C, 3:D, 4:A

Answer Key

- 1. d While all elements are pure substances, not all pure substances are elements. This is because while elements are a type of pure substance made up of only one type of atom, pure substances can also include compounds made up of two or more types of atoms.
- 2. b The sulfur powder will sublime, leaving the iron filings behind. Sublimation is the process in which a solid directly converts into a gas without going through the liquid state. In this experiment, when heated, the sulfur powder will sublime, turning into a gas and leaving the iron filings behind.
- **3.** b Compounds are substances made up of two or more elements chemically combined. Sugar is a compound composed of carbon, hydrogen, and oxygen atoms. Glass is a mixture of compounds but can be considered a compound itself. Carbon dioxide is a compound made up of carbon and oxygen atoms, and oxygen gas (O₂) is a molecule with two atoms and can also be considered a compound.
- **4.** a In the of distillation process, the mixture is heated (boiling) to vaporize the ethanol. The ethanol vapours are then cooled (condensation) to convert them back into a liquid state. Finally, the liquid ethanol is collected.
- **5.** d Filtration: Separation of solid particles from a liquid or gas by passing it through a porous medium.

Magnetic separation: Separation of magnetic substances from non-magnetic substances. Winnowing: Separation of heavier or lighter components of a mixture by wind or blowing air. Threshing: Separation of grains from stalks by beating or crushing.

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